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## Tracing TiO<sub>2</sub> nanoparticles in a hard rock aquifer

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The increasing use of TiO<sub>2</sub> nanoparticles (NPs) is inevitably leading to its introduction into different environmental compartments, including soils, surface water and groundwater. Where groundwater is widely used for drinking water, any contamination could have an adverse impact on human health as already shown for TiO<sub>2</sub> NPs on living organisms and human lymphocytes. Consequently, the potential exposure of TiO<sub>2</sub> NPs through groundwater contamination needs to be studied and assessed.

The small-scale tracer experiment conducted in a schist aquifer constitutes an attempt to understand the physical and chemical processes occurring if TiO<sub>2</sub> NPs happen to penetrate in a hard rock aquifer. The schist aquifer can be represented by a dual-porosity medium constituted of fractures with low longitudinal dispersivity in which water and solute transport is relatively fast, and of a network of smaller fissures with high longitudinal dispersivity in which transport is slower.

One of the major results is that TiO<sub>2</sub> NPs were mobile and transferred in a heterogeneous schist aquifer through its conductive zones under aggregate forms. NPs do not penetrate the entire poral network during transfer, especially the one with very low porosity, due to electrostatic interactions between NPs aggregates and rock. During their transfer, a large amount of NPs was also deposited onto the rock surface, i.e. affected by sedimentation, clogging and adsorption. Weakly negatively charged NPs may be adsorbed onto the rock surface and especially on the positively charged iron oxi-hydroxides that coat the main pathways due to natural denitrification process. NPs aggregation is reversible due to the mineralogical heterogeneity of the schist and is likely to evolve rapidly after injection and during transport according to the environmental conditions.

Any modification of the aquifer chemical conditions is likely to impact the groundwater pH, the nitrate content and the denitrification process that controls the precipitation of iron oxides in the flow pathways, and thus change NPs aggregation and sorption.